

Physical Chemistry 1 Acs Exam Study Guide

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Physical Chemistry 1 Acs Exam

ACS Standard Physical Chem Exam Taken Tuesday by 21 ...

1) if C_v constant For ideal gas the above is always true (even if volume changes) $H = U + PV$ $\Delta H = \Delta U + P \Delta V$ For $P = P_{ext} = \text{constant}$ and PV work only: $\Delta H = q = \Delta H = C_p (T_2 - T_1)$ if C_p constant For ideal gas the above is always true (even if pressure changes) for P_{ext} constant: $w_{pv} = -P_{ext} (V_2 - V_1)$; if

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CHEM 542 - Spring 2016 Physical Chemistry: Quantum ...

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