

# Stoichiometry Homework Packet On Moles

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## Stoichiometry Homework Packet On Moles

### PracticePacket((Unit6: Moles(&(Stoichiometry

PRACTICEPACKET:((Unit(6Moles(&(Stoichiometry((7(wwwmrpalermocom(Usetheformula!below!to!answer!questions!8?13! Fe 2O  
3!+3CO!!2Fe+2CO 2! (8 If(300(moles(of(Iron

### Unit 7: Stoichiometry Homework Packet (85 points)

Unit 7: Stoichiometry Homework Packet (85 points) goldchemistrywordpresscom Page 2 Grams to Moles or Moles to Grams: Answer the following questions Be sure to show all your work and your units (pg 390-398) 4 The decomposition of potassium chlorate, KClO<sub>3</sub>, is used as a source of oxygen in the laboratory How many moles of potassium

### Stoichiometry Unit 8 1 Mole Relationships Answers

Add these notes to you notes packet on page 3 2 Complete page 8 in your practice packet (The second half of the page involves 2 steps This will be tomorrow's lesson and I have already posted that video) 3 Complete and submit Homework #4 on google classroom Unit 8 - Moles and Stoichiometry  
- ...

### Early Booklet E.C.: / 2 Total HW Points Unit 7 - The Mole ...

How many moles of oxygen atoms are contained in each compound? 1 250 mol of KMnO<sub>4</sub> 2 459 mol of CO<sub>2</sub> 3 What is the molar mass of sodium nitrate, NaNO<sub>3</sub>? 4 Garlic Determine the molar mass of allyl sulfide, the compound responsible for the smell of garlic The chemical formula of allyl sulfide is (C<sub>3</sub>H<sub>5</sub>)<sub>2</sub>S How many moles are in 1000

### Stoichiometry - Practice Problems

Homework Set 1: 1) 2C<sub>3</sub>H<sub>7</sub>OH + 9O<sub>2</sub> → 6CO<sub>2</sub> + 8H<sub>2</sub>O a) How many moles of O<sub>2</sub> are required to react with 58 moles of C<sub>3</sub>H<sub>7</sub>OH? b) How many moles of CO<sub>2</sub> are required to react with 17 moles of O<sub>2</sub>? c) How many grams of CO<sub>2</sub> would be required to react with 78 moles of H<sub>2</sub>O? d)

How many grams of  $C_3H_7OH$  are needed to produce 0.45 moles

### Chemistry Stoichiometry Packet Answers

Stoichiometry Packet Answers Chapter 12 Stoichiometry Packet Answer Key fullexams com Unit 6 Reactions and Stoichiometry STUDENT REVIEW PACKET ANSWER KEY PBworks Classroom Resources Sweet Stoichiometry Reactions AACT Classwork and Homework Handouts Stoichiometry Review Packet Stoichiometry and the Mole PracticePacket Unit6 Moles amp

### Chemistry B Moles Packet Answers

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### mc06se cFMsr i-vi

(b) relative number of moles of reactants and products (c) number of atoms of each element in each compound in a reaction (d) number of valence electrons involved in a reaction 2 d Which of the following would not be studied within the topic of stoichiometry? (a) the mole ratio of Al to Cl in the compound aluminum chloride

### Unit 6: Reactions and Stoichiometry

Unit 6: Reactions and Stoichiometry (Link to Prentice Hall Text: Chapters 7, 8 & 9) Name: \_\_\_\_ Date Due Assignments Page Number: Problem Numbers Assignment 1: Interpreting Equations and Balancing 232: 32, 39 Assignment 2: Predicting Reaction Products, Net Ionic Equations

### stoichiometry 1 worksheet and key - Saddleback College

$2 KClO_3 \rightarrow 2 KCl + 3 O_2$  10 How many grams of  $O_2$  will be formed from 376 grams of  $KClO_3$ ? 11 How many grams of  $KClO_3$  are needed to make 300 grams of  $KCl$ ? 12 How many grams of  $KCl$  will be formed from 273 g of  $KClO_3$ ?  $4 Fe + 3 O_2 \rightarrow 2 Fe_2O_3$  13 How many grams of  $Fe_2O_3$  are produced when 427 grams of  $Fe$  is reacted? 14 How many grams of  $Fe_2O_3$  are produced when ...

### Stoichiometry Unit 8 1 Mole Relationships Answers

Watch the video on mole and liter conversions Add these notes to you notes packet on page 3 2 Complete page 8 in your practice packet (The second half of the page involves 2 steps This will be tomorrow's lesson and I have already posted that video) 3 Complete and submit Homework #4 on google classroom Unit 8 - Moles and Stoichiometry

### Teacher: Subject: Chemistry-Stoichiometry Unit Grade ...

(given moles, particles, mass, or volume and ending with moles, particles, mass, or volume of the desired substance) By the end of class students will be able to: 1 Identify the given and unknown in a stoichiometry word problem 2 Correctly label a balanced chemical equation with the given and unknown information 3

### Cerritos College - Norwalk, CA

WORKSHEET- STOICHIOMETRY SET A: (Time required, 1 hour) A compound with the formula,  $BxH_2O_3$ , contains 36.14 % by mass oxygen What is the value of the 1) Ans:  $x = 6$  H C O 14 - 22 20 36,)  $14 2\ddot{Y}q$  A mixture of cobalt(I) oxide and cobalt(III) oxide contains 32.50 % ...

### Unit 11 Packet - Page 1 of 14 Honors Chemistry - Unit 11

$P = atm$   $V = L$   $T = K$   $n = moles$   $R = gas\ constant\ 0.0821\ L\cdot atm/mol\cdot K$  (memorize) Example: What is the pressure exerted by a 120 g sample of Nitrogen gas ( $N_2$ ) in a 100 L container at 25 °C? Practice Ideal Gas Law Worksheet: 1 - 4 (page 8 in packet) Gas Stoichiometry Molar Volume - 1

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mol of any gas at STP has a volume of 22.4 L

### **Chemistry Stoichiometry Packet Answers**

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### **Chemistry B Moles Packet Answers**

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### **Date. FCHAPJ REV[EW.**

Stoichiometry SHORT ANSWER Answer the following questions in the space provided 1 b The coefficients in a chemical equation represent the (a masses in grams of all reactants and products (h) relative number of moles of reactants and products (c) number of atoms of each element in each compound a reaction